# $v$-Triazolines. Part 42. ${ }^{1}$ Study on the reactivity of 4,5-dihydro-1-(6-methyl-2-oxo-2H-pyran-4-yl)-5-morpholino-v-triazoles. Synthetic approach to pyrano[4,3-b]pyrrol-4(1H)-ones 

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Pyrolysis of 4-aryl-5-morpholino-4,5-dihydrotriazoles $\mathbf{3}$ affords two products: pyrano[4,3-b]pyrrol-4(1H)-ones $\mathbf{4}$ and arylacetamidines $\mathbf{5}$. The reaction mechanism of this transformation is discussed and reaction conditions optimized to enhance the formation of pyrrole-fused pyran-2-one derivatives 4. 2-Aminoaziridines are considered to be key intermediates in this transformation.

2 H -Pyran-2-ones are of interest as synthons for organic synthesis, particularly since many derivatives, isolated from natural sources, exhibit remarkable biological properties. ${ }^{2}$ Renewed interest comes from the discovery of a new type of non-peptidic HIV-protease inhibitor containing this nucleus. ${ }^{3}$

In 2 H -pyranones the $\mathrm{C}-3$ position is nucleophilic, presenting the characteristic reactivity of enols, and electrophiles react at $\mathrm{C}-3$ with conservation of the pyrone ring. ${ }^{4}$
Our research group has been studying transformations and uses of 5 -amino-4,5-dihydro- $v$-triazoles, mainly as precursors of substituted amidines, synthons from which N-heterocycles can be derived. ${ }^{5}$ It is well known that 5 -amino- 4,5 -dihydro- $v$ triazoles readily aromatize to triazoles ${ }^{6 a}$ by heat or under the action of acid or base by elimination of a molecule of amine. The 5-amino-4,5-dihydro- $v$-triazoles give rise to amidines when the R group on $\mathrm{N}-1$ is highly electron withdrawing and the $5-$ amino group is tertiary. ${ }^{6 b}$ In a previous paper ${ }^{7}$ we pointed out that thermolysis can induce 4,5 -dihydro- $v$-triazole ring cleavage (Scheme 1). The zwitterionic intermediate, produced by ringchain tautomerism of the triazoline ring, undergoes two different reactions: direct formation of the amidine by $\mathrm{N}_{2}$ loss and simultaneous hydride transfer, or alternatively ring contraction into the 2-aminoaziridine, associated with $\mathrm{N}_{2}$ loss. We also observed the formation of labile diaminoethylenic derivatives, whose formation has been rationalized by invoking the presence of an aziridine intermediate. Cleavage of the aziridine bond between N and the C atom bearing the amino residue, through its iminium form, explained the ethylenic intermediates.

These results prompted us to synthesize 5 -amino- 4,5 -di-hydro- $v$-triazoles bearing a 2 H -pyran- 2 -one group at $\mathrm{N}-1$, and to explore the possibility of linking both the reactivity of 2-aminoaziridines and the nucleophilic features of $\mathrm{C}-3$ in the 2 H -pyran-2-one nucleus.

Suitable conditions were sought to achieve transformation of the intermediate aminoaziridine pyrrole-fused pyran-2-one derivatives. Little chemistry of this ring system has been investigated ${ }^{8}$ although some substituted derivatives have been prepared from pyrrol-2-ylacetic acid derivatives ${ }^{9}$ by a multistep sequence. A synthetic pathway involving 5 -amino-4,5-dihydro- $v$-triazoles has never been described.


Scheme 1

## Results and discussion

4,5-Dihydro- $\boldsymbol{v}$-triazoles $\mathbf{3 a - d , f , g}$ were easily obtained from cycloaddition of azide $\mathbf{1}$ and appropriate enamines $\mathbf{2 a - d , f , g}$ in benzene while 3 e was prepared by one-pot procedure ${ }^{10}$ from propionaldehyde, 4-azido-6-methyl-2H-pyran-2-one $\mathbf{1}$ and morpholine without isolation of the intermediate enamine $2 \mathbf{e}$ (Scheme 2). Satisfactory yields were obtained and the structures of products, including their trans configuration, ${ }^{11}$ were established from ${ }^{1} \mathrm{H}$ NMR data (see Experimental section).

Compounds $\mathbf{3 a - g}$ were thermolyzed by heating them in boiling toluene until complete consumption of the starting


Scheme $2{ }^{a}$ 2e not isolated, but obtained in situ: see Experimental section, preparation of compound $\mathbf{3 e}$.
compound (about 1-2 hours). Starting from 3a-d,f,g a mixture of pyrrole-fused pyran-2-ones $\mathbf{4 a - d , f , g}$ and amidines $\mathbf{5 a - d}, \mathbf{f}, \mathbf{g}$, respectively, was formed. Dihydro- $\nu$-triazole 3e gave only amidine 5e (Scheme 3).


Scheme $3{ }^{a} \mathbf{4 e}$ was obtained in toluene but only in the presence of $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ (Tables 1 and 2).

Spectroscopic data established the structure of tertiary amidines $5 \mathbf{a - g}$ and of 2-aryl-6-methylpyrano[4,3-b]pyrrol$4(1 H)$-ones $\mathbf{4 a - d , f , g}$. Assignment of protons and carbons in ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ spectra of amidines $\mathbf{5 a - g}$ was made on the basis of HETCOR and COLOC experiments, and data related to the pyran-2-one nucleus were in agreement with the literature. ${ }^{12}$ All amidines 5 (in $\mathrm{CDCl}_{3}$ ) show a typical pattern for the signal of the $\mathrm{C}-3$ proton in the range $\delta 5.15-5.29$ and for the $\mathrm{C}-5$ proton in the range $\delta 5.58-5.66$. A NOESY experiment, performed for amidine 5a, demonstrated correlation between the singlet at $\delta 3.8\left(\mathrm{CH}_{2} \mathrm{Ph}\right)$ and the singlet at $\delta 5.29$ (pyran-2-one $3-\mathrm{H}$ ), supporting the assigned ( $E$ ) configuration of the amidine double bond.

All pyrano[4,3-b]pyrrol-4-ones $\mathbf{4 a - d}, \mathbf{f}, \mathbf{g}$ show typical proton signals (in DMSO- $\mathrm{d}_{6}$ ) due to the $\mathrm{C}-7$ and $\mathrm{C}-3$ protons at $\delta \approx 6.5$ and $\delta 6.87-7.07$, respectively. The complete assignment of the peaks in the ${ }^{13} \mathrm{C}$ NMR spectrum was facilitated by 2D NMR spectroscopy performed on the pyrano[4,3-b]pyrrol-4(1H)-one 4a: HETCOR experiments allowed attribution of the CH-7 signals ( $\delta 6.48$ and $\delta_{\mathrm{C}} 96$ ) as well as the CH- $3 \operatorname{signal}$ ( $\delta 7.01$ and
$\delta_{\mathrm{C}} \approx 103$ ), quaternary carbons being assigned from the results of a COLOC experiment. For all compounds 4 proton/carbon resonances are reported in the Experimental section.

Conclusive structural assignment of compounds $\mathbf{4 a - d} \mathbf{d} \mathbf{f}, \mathbf{g}$ was obtained by NOESY experiments ( $\mathrm{DMSO}-\mathrm{d}_{6}$ ) on $\mathbf{4 a}$, which displayed clear correlations between $\mathrm{H}-7$ and the $\mathrm{CH}_{3}$ linked to C-6. A positive Overhauser effect confirmed also the spatial proximity of the NH group to $\mathrm{H}-7$ as well as to the ortho hydrogens of the C-2 phenyl group.

The formation of amidines $\mathbf{5}$ is expected ${ }^{6 b}$ according to the general rearrangement path of 5-amino-4,5-dihydrotriazoles as indicated in Scheme 1. The production of fused heterocycles 4 can be rationalized as occurring via aminoaziridine $\mathbf{C}$ (Scheme 4). The previously observed ${ }^{7}$ bond cleavage between


Scheme 4
the N atom and the C atom linked to the morpholine residue forms $\mathbf{D}$, where nucleophilic attack of $\mathbf{C}-3$ of the pyran-2-one on the iminium carbon occurs to give a dihydropyrano[4,3-b]-pyrrol-4-one intermediate, which, by amine loss, turns into the corresponding pyrano[4,3-b]pyrrol-4(1H)-one 4.

The reaction was studied in more detail, both to confirm that bicycles $\mathbf{4}$ do not arise from the amidines 5 , and to determine those factors that would enhance the yield of the pyrrole-fused pyran-2-ones 4. In a first experiment amidine $\mathbf{5 a}$ was refluxed in toluene for 60 min , under the same conditions adopted for the decomposition of 4,5 -dihydro- $v$-triazole $3 \mathbf{3}$. The ${ }^{1} \mathrm{H}$ NMR spectrum of the crude mixture showed only the signals of amidine $\mathbf{5 a}$, thus confirming the independence of paths leading to 4 and 5, respectively.

In order to increase the yield of compounds 4, 4,5-dihydro- $v$ triazoles $\mathbf{3 a - g}$ were treated under two different conditions: i) refluxing in propan-1-ol until disappearance of starting material was complete; ii) boiling in dry toluene with an equimolar amount of $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ for the same time. Propan-1-ol was selected on account of its suitable boiling point. It was

Table 1 Ratios of $\mathbf{4}$ and $\mathbf{5}$ estimated by ${ }^{1} \mathrm{H}$ NMR spectroscopy

| Compound <br> $\mathbf{3}$ | Toluene | $\operatorname{Pr}^{n} \mathrm{OH}$ | Toluene, <br> $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ |
| :--- | :--- | :--- | :--- |
| $\mathbf{3 a}$ | $\mathbf{4 a}: \mathbf{5 a}(33: \mathbf{6 7 )}$ | $\mathbf{4 a}: \mathbf{5 a}(18: 82)$ | $\mathbf{4 a}: \mathbf{5 a}(70: 30)$ |
| $\mathbf{3 b}$ | $\mathbf{4 b}: \mathbf{5 b}(38: 62)$ | $\mathbf{4 b}: \mathbf{5 b}(34: 66)$ | $\mathbf{4 b}: \mathbf{5 b}(88: 12)$ |
| $\mathbf{3 c}$ | $\mathbf{4 c}: \mathbf{5 c}(37: 63)$ | $\mathbf{4 c}: \mathbf{5 c}(66: 34)$ | $\mathbf{4 c}: \mathbf{5 c}(82: 18)$ |
| $\mathbf{3 d}$ | $\mathbf{4 d}: \mathbf{5 d}(44: 56)$ | $\mathbf{4 d}: \mathbf{5 d}(19: 81)$ | $\mathbf{4 d}: \mathbf{5 d}(94: 6)$ |
| $\mathbf{3 e}$ | $\mathbf{4 e}: \mathbf{5 e}(0: 100)$ | $\mathbf{4 e}: \mathbf{5 e}(0: 100)$ | $\mathbf{4 e}: \mathbf{5 e}(25: 75)$ |
| $\mathbf{3 f}$ | $\mathbf{4 f}: \mathbf{5 f}(48: 52)$ | $\mathbf{4 f}: \mathbf{5 f}(28: 72)$ | $\mathbf{4 f}: \mathbf{5 f}(83: 17)$ |
| $\mathbf{3 g}$ | $\mathbf{4 g}: \mathbf{5 g}(32: 68)$ | $\mathbf{4 g}: \mathbf{5 g}(22: 78)$ | $\mathbf{4 g}: \mathbf{5 g}(63: 37)$ |

expected that this polar and protic solvent might influence the equilibrium between the conformers $\mathbf{A}$ and $\mathbf{B}$ previously seen in Scheme 1. In this reaction setting we were expecting the dipolar form $\mathbf{A}$ to prevail and, from $\mathrm{N}_{2}$ loss, the amount of amidines 5 to increase.

Dry toluene and $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ were used in attempts to stabilize the iminium intermediate $\mathbf{D}$ in Scheme 4. It is well known that aziridines with electron-withdrawing groups, ${ }^{13}$ such as alkylsulfonyl, acyl, alkoxycarbonyl, and in our case the unsaturated lactone pyran-2-one linked to the N atom, conjugatively stabilize the negative charge that develops on the nitrogen in the ring-opening transition state. In this context (see Scheme 4) $\mathrm{BF}_{3}$ could bond to the anionic nitrogen, in iminium form $\mathbf{D}$, or conjugatively to the enolate oxygen atom and hence promote the formation of 4 . Table 1 collects the results obtained under three different conditions. The ratio $\mathbf{4}: 5$ in the product mixture was determined by ${ }^{1} \mathrm{H}$ NMR analysis.

The data of Table 1 confirm the correctness of the above choices and give support to the formulated mechanistic hypothesis, that the achievement of heterocycles 4 and the amidines 5 comes from two competitive pathways. An effective increase in the yield of amidines 5 was seen when the reaction was performed propan-1-ol, while the addition of the Lewis acid catalyst brought about an increase of the yield of pyrrolefused pyran-2-one derivatives 4, suggesting that the participation of polar solvents and/or complexing catalyst can act on the equilibrium between $\mathbf{A}$ and $\mathbf{B}$, favouring the hydride transfer or the formation of the aziridine intermediate.

In conclusion, variations in the reaction medium controlled the transformation of 4,5-dihydro- $v$-triazoles 3 and facilitated our aim of linking the reactivity of 2 -aminoaziridines and the nucleophilic reactivity of $\mathrm{C}-3$ in the 2 H -pyranone nucleus.

## Experimental

Mps were determined using a Büchi 510 (capillary) or an Electrothermal 9100 apparatus and are uncorrected. IR spectra were measured using a JASCO IR Report 100 instrument. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra (tetramethylsilane as internal standard) were recorded with EM Varian Gemini 200, Bruker AC 200 and Bruker Avance 300 Spectrometers. $J$-Values are given in Hz for solutions in $\mathrm{CDCl}_{3}$ or $\mathrm{DMSO}-d_{6}$. Mass spectral data were obtained on a Varian MAT 1H COS 50 instrument using electron-impact ionization techniques at 70 eV . Column chromatography was performed on Kieselgel 60 (Merck), $0.063-0.200 \mathrm{~mm}$, and elution with cyclohexane-ethyl acetate (ratios indicated in Experimental section).

## Materials

Azide $\mathbf{1}^{14}$ and enamines $\mathbf{2 a - c},{ }^{\mathbf{1 5}} \mathbf{2 f},{ }^{\mathbf{1 5}} \mathbf{2 d},{ }^{16} \mathbf{2 g}{ }^{\mathbf{1 7}}$ have already been described.

## 4,5-Dihydro-1-(6-methyl-2-oxo-2H-pyran-4-yl)-5-morpholino-1,2,3-triazoles 3a-d,f,g

Typical procedure. Azide $1(3.0 \mathrm{~g}, 20 \mathrm{mmol})$ was dissolved in benzene $(20 \mathrm{ml})$ and an equimolar amount of an enamine $\mathbf{2 a}, \mathbf{b}$,
$\mathbf{d}, \mathbf{f}, \mathbf{g}$, dissolved in benzene ( 20 ml ), was added under stirring at room temperature. Enamine 2c solution in benzene was added dropwise to the stirred azide solution at $0-5^{\circ} \mathrm{C}$. As soon as the starting azide had disappeared (TLC, cyclohexane-ethyl acetate $3: 7$ ), the solvent was removed under reduced pressure and the crude product was crystallized from an appropriate solvent.

4-(4,5-Dihydro-5-morpholino-4-phenyl-1 H -1,2,3-triazol-1-yl)-6-methyl-2H-pyran-2-one 3a. Reaction time 2 h ; yield 6.5 g ( $96 \%$ ); white plates from $\operatorname{Pr}^{\mathrm{i}}{ }_{2} \mathrm{O}$; mp $85^{\circ} \mathrm{C}$ (decomp.); IR (Nujol) $v_{\text {max }} 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 2.30(3 \mathrm{H}$, $\mathrm{s}, \mathrm{Me}), 2.39-2.47\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 3.64-3.72(4 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 4.62(1 \mathrm{H}, \mathrm{d}, J 3.4,5-\mathrm{H}), 5.61(1 \mathrm{H}, \mathrm{d}, J 3.4$, $4-\mathrm{H}), 5.72(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}$ pyranone $), 6.78(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}$ pyranone $)$, 6.95-7.42 (5H, m, ArH) (Calc. for $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{~N}_{4} \mathrm{O}_{3}: \mathrm{C}, 63.52$; H, 5.92; N, 16.46. Found: C, 63.69; H, 6.03; N, 16.18\%).

## 4-(4,5-Dihydro-5-morpholino-4-p-tolyl-4-1 $\mathbf{H - 1 , 2 , 3 - t r i a z o l - 1 - ~}$

 yl)-6-methyl-2H-pyran-2-one 3b. Reaction time 2 h ; yield 5.7 g ( $81 \%$ ); white plates from $\operatorname{Pr}_{2}{ }^{\mathrm{O}} \mathrm{O}$; mp $108^{\circ} \mathrm{C}$ (decomp.); IR (Nujol) $v_{\max } 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $\delta 2.29(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.33(3 \mathrm{H}, \mathrm{s}, p$-tolyl), $2.35-2.44(4 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 3.63-3.69\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 4.58(1 \mathrm{H}, \mathrm{d}, J 3.4$, $5-\mathrm{H}), 5.57(1 \mathrm{H}, \mathrm{d}, J 3.4,4-\mathrm{H}), 5.71(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}$ pyranone $), 6.78$ $(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}$ pyranone $), 6.87$ and $7.17(4 \mathrm{H}, \mathrm{dd}, \mathrm{AB}$ system, $J 8.0$, ArH (Calc. for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{3}$ : C, 64.39; $\mathrm{H}, 6.26 ; \mathrm{N}, 15.81$. Found: C, 64.6; H, 6.31; N, 15.55\%).4-[4,5-Dihydro-4-(4-Methoxyphenyl)-5-morpholino-1H-1,2,3-triazol-1-yl]-6-methyl-2H-pyran-2-one 3c. Reaction time 2 h ; yield $4.7 \mathrm{~g}(64 \%)$; white plates from $\operatorname{Pr}_{2}^{\mathrm{i}} \mathrm{O} ; \mathrm{mp} 65-66^{\circ} \mathrm{C}$ (decomp.); IR (Nujol) $v_{\max } 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (200 $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 2.30(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.38-2.47(4 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 3.63-3.70\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 3.80(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, $4.58(1 \mathrm{H}, \mathrm{d}, J 3.4,5-\mathrm{H}), 5.56(1 \mathrm{H}, \mathrm{d}, J 3.4,4-\mathrm{H}), 5.71(1 \mathrm{H}, \mathrm{s}$, 3-H pyranone), $6.79(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}$ pyranone $), 6.88-6.98(4 \mathrm{H}$, m, ArH ) (Calc. for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{~N}_{4} \mathrm{O}_{4}$ : C, $61.61 ; \mathrm{H}, 5.99 ; \mathrm{N}, 15.13$. Found: C, 61.84; H, 6.12; N, 14.97\%).

4-[4-(4-Chlorophenyl)-4,5-dihydro-5-morpholino-1 $\mathrm{H}-1,2,3-$ triazol-1-yl]-6-methyl-2H-pyran-2-one 3d. Reaction time 24 h ; yield $5.5 \mathrm{~g}(73 \%)$; white plates from benzene; mp $107^{\circ} \mathrm{C}$ (decomp.); IR (Nujol) $v_{\max } 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (200 $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 2.30(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.38-2.48(4 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 3.64-3.75\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 4.58(1 \mathrm{H}, \mathrm{d}, J 3.5$, $5-\mathrm{H}), 5.58(1 \mathrm{H}, \mathrm{d}, J 3.5,4-\mathrm{H}), 5.71(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}$ pyranone $)$, $6.75(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}$ pyranone $), 6.94$ and $7.36(4 \mathrm{H}, \mathrm{dd}, \mathrm{AB}$ system, $J 8.5, \mathrm{ArH})\left(\right.$ Calc. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{ClN}_{4} \mathrm{O}_{3}: \mathrm{C}, 57.68 ; \mathrm{H}, 5.11 ; \mathrm{N}$, 14.95. Found: C, 57.89 H, $5.18 ;$ N, $14.68 \%$ ).

4-[4-(4-Bromophenyl)-4,5-dihydro-5-morpholino-1 $\mathbf{H - 1 , 2 , 3 -}$ triazol-1-yl]-6-methyl-2H-pyran-2-one 3f. Reaction time 2 h ; yield $5.4 \mathrm{~g}(64 \%)$; cream plates from ethanol; mp $115^{\circ} \mathrm{C}$ (decomp.); IR (Nujol) $v_{\max } 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (200 $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 2.32(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.36-2.47(4 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 3.64-3.75\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 4.60(1 \mathrm{H}, \mathrm{d}, J 3.6$, $5-\mathrm{H}), 5.58(1 \mathrm{H}, \mathrm{d}, J 3.6,4-\mathrm{H}), 5.73(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}$ pyranone $)$, $6.79(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}$ pyranone $), 6.89$ and $7.53(4 \mathrm{H}, \mathrm{dd}, \mathrm{AB}$ system, $J$ 8.4, ArH) (Calc. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{BrN}_{4} \mathrm{O}_{3}: \mathrm{C}, 51.56 ; \mathrm{H}, 4.57 ; \mathrm{N}$, 13.36. Found: C, 51.63 ; H, 4.81; N, 13.07\%).

4-[4-(4-Fluorophenyl)-4,5-dihydro-5-morpholino-1 $\mathbf{H - 1 , 2 , 3 -}$ triazol-1-yl]-6-methyl-2H-pyran-2-one 3g. Reaction time 3 h ; yield 4.1 g ( $57 \%$ ); cream plates from ethanol; mp $118^{\circ} \mathrm{C}$ (decomp.); IR (Nujol) $v_{\max } 1690(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (200 $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 2.33(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.40-2.47(4 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 3.68-3.75\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 4.60(1 \mathrm{H}, \mathrm{d}, J 3.6$, $5-\mathrm{H}), 5.60(1 \mathrm{H}, \mathrm{d}, J 3.6,4-\mathrm{H}), 5.74(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}$ pyranone $), 6.80$ $(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}$ pyranone $), 6.90-7.16(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH})($ Calc. for

Table 2 Preparation of pyranol[4,3-b]pyrrol-4(1H)-ones $\mathbf{4}$ or amidines $\mathbf{5}$ from 5-amino-4,5-dihydro- $\nu$-triazoles $\mathbf{3}$

| Entry | R | Method ${ }^{\text {a }}$ | Reaction time ( $t / \mathrm{min}$ ) | $T /{ }^{\circ} \mathrm{C}$ | Yield of isolated products (\%) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3a | Ph | A | 60 | 111 | $\mathbf{4 a}$ (21) 5a (55) |
| 3 a | Ph | B | 120 | 97 | $\mathbf{4 a}$ (13) 5a (70) |
| 3 a | Ph | C | 60 | 111 | $\mathbf{4 a}$ (58) 5a (18) |
| 3b | $4-\mathrm{MeC}_{6} \mathrm{H}_{4}$ | A | 90 | 111 | $\mathbf{4 b}$ (24) $\mathbf{5 b}$ (46) |
| 3b | $4-\mathrm{MeC}_{6} \mathrm{H}_{4}$ | B | 90 | 97 | $\mathbf{4 b}$ (22) 5b (56) |
| 3b | $4-\mathrm{MeC}_{6} \mathrm{H}_{4}$ | C | 90 | 111 | $\mathbf{4 b}$ (77) 5b (7) |
| 3 c | $4-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | A | 90 | 111 | 4c (25) 5c (48) |
| 3 c | $4-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | B | 120 | 97 | 4c (53) 5c (20) |
| 3 c | $4-\mathrm{MeOC}_{6} \mathrm{H}_{4}$ | C | 90 | 111 | 4c (70) 5c (13) |
| 3d | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | A | 100 | 111 | 4d (36) 5d (47) |
| 3d | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | B | 80 | 97 | 4d (11) 5d (68) |
| 3d | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | C | 120 | 111 | 4d (88) 5d (2) |
| 3 e | Me | A | 120 | 111 | 5e (86) |
| 3 e | Me | B | 120 | 97 | 5e (88) |
| 3 e | Me | C | 120 | 111 | 4e (16) 5e (58) |
| 3 f | $4-\mathrm{BrC}_{6} \mathrm{H}_{4}$ | A | 120 | 111 | $\mathbf{4 f}$ (36) $\mathbf{5 f}$ (43) |
| 3 f | $4-\mathrm{BrC}_{6} \mathrm{H}_{4}$ | B | 120 | 97 | $\mathbf{4 f}$ (15) $\mathbf{5 f}$ (60) |
| 3 f | $4-\mathrm{BrC}_{6} \mathrm{H}_{4}$ | C | 120 | 111 | $\mathbf{4 f}$ (65) $\mathbf{5 f}$ (10) |
| 3 g | $4-\mathrm{FC}_{6} \mathrm{H}_{4}$ | A | 120 | 111 | $\mathbf{4 g}$ (22) $\mathbf{5 g}(56)$ |
| 3 g | $4-\mathrm{FC}_{6} \mathrm{H}_{4}$ | B | 120 | 97 | $\mathbf{4 g}$ (13) $\mathbf{5 g}$ (64) |
| 3g | $4-\mathrm{FC}_{6} \mathrm{H}_{4}$ | C | 120 | 111 | $\mathbf{4 g}$ (45) 5g (30) |
| ${ }^{a}$ Solvents: A, toluene; B, propan-1-ol; C, toluene, $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$. |  |  |  |  |  |

$\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{FN}_{4} \mathrm{O}_{3}: \mathrm{C}, 60.33 ; \mathrm{H}, 5.34 ; \mathrm{N}, 15.63$. Found: C, 60.46; $\mathrm{H}, 5.41$; N, $15.35 \%$ ).

Preparation of 4-(4,5-dihydro-4-methyl-5-morpholino-1 $\mathbf{H - 1 , 2 , 3 -}$ triazol-1-yl)-6-methyl-2H-pyran-2-one 3 e
A benzene solution ( 30 ml ) of morpholine $(2.61 \mathrm{~g}, 30 \mathrm{mmol})$ was added dropwise to a stirred solution of azide $1(4.5 \mathrm{~g}$, $30 \mathrm{mmol})$ and propanal $(1.74 \mathrm{~g}, 30 \mathrm{mmol})$ in benzene $(30 \mathrm{ml})$ at rt until the starting azide had disappeared (ca. 20 h ) (TLC cyclohexane-EtOAc, 1:9). The solution was dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and the filtrate was evaporated under reduced pressure. The residue was crystallized from $\mathrm{Pr}_{2}^{\mathrm{i}} \mathrm{O}$ to afford pure 3e. ( $5.6 \mathrm{~g}, 67 \%$ ), as white needles; $\mathrm{mp} 113{ }^{\circ} \mathrm{C}$ (decomp.); IR (Nujol) $v_{\text {max }} 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $\delta$ $1.28(3 \mathrm{H}, \mathrm{d}, J 7.2$, 4-Me triazoline), $2.28(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 2.30-2.38$ $\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 3.61-3.68\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 4.31(1 \mathrm{H}$, d, $J 3.1,5-\mathrm{H}), 4.58(1 \mathrm{H}, \mathrm{dq}, J 3.1$ and $7.2,4-\mathrm{H}), 5.70(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}$ pyranone), $6.72\left(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}\right.$ pyranone) (Calc. for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{~N}_{4} \mathrm{O}_{3}$ : C, 56.10; H, 6.52; N, 20.13. Found: C, $56.22 ; \mathrm{H}, 6.64 ;$ N, 20.01\%).

## Thermal behaviour of dihydrotriazoles 3a-g. Method A

Typical procedure for the synthesis of compounds $4 a-\mathbf{d , f}, \mathrm{g}$ and $\mathbf{5 a - g}$. Dihydrotriazoles $\mathbf{3 a - g}(15 \mathrm{mmol})$ dissolved in toluene $(70 \mathrm{ml})$ were heated under reflux for times indicated in Table 2, progress of the reaction being followed by TLC. After disappearance of the starting material the solvent was removed in vacuo and a small amount of the residue was dissolved in DMSO- $d_{6}$ and the ratio of 4 to 5 was determined by ${ }^{1} \mathrm{H}$ NMR analysis (see Table 1). The crude residue was taken up with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and slowly and partially deposited a compound $\mathbf{4 a}-$ $\mathbf{d , f , g}$ as a crystalline product, which was filtered off and recrystallized as indicated below. The left over mixture was chromatographed (cyclohexane-ethyl acetate $3: 7$ ) to give a first fraction containing a pyrano[4,3-b]pyrrol-4-one 4a-d,f,g and a second fraction containing an amidine $\mathbf{5 a - d}, \mathbf{f}, \mathbf{g}$. Dihydrotriazole $\mathbf{3 e}$ afforded only amidine $\mathbf{5 e}$, isolated by crystallization from $\operatorname{Pr}^{i}{ }_{2} \mathrm{O}$. Pyrrole-fused pyran-2-one $\mathbf{4 e}$ was isolated only from reaction with $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ (see below). Reaction times of dihydrotriazoles $\mathbf{3}$ and yields of isolated products $\mathbf{4}$ and 5 are collected in Table 2.

6-Methyl-2-phenylpyrano[4,3-b]pyrrol-4(1H)-one 4a. White
plates from $\operatorname{Pr}_{2}^{\mathrm{i}} \mathrm{O} ; \operatorname{mp} 223{ }^{\circ} \mathrm{C}$; IR (Nujol) $v_{\max } 3250(\mathrm{NH}), 1675$ $(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz} ; \mathrm{DMSO}-d_{6}$ ) $\delta 2.26(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Me}), 6.48(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 7.01(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 7.19-7.82(5 \mathrm{H}, \mathrm{m}$, ArH), 12.2 ( 1 H , br s, NH exchangeable); ${ }^{13} \mathrm{C}$ NMR ( 75 MHz ; DMSO- $d_{6}$ ) $\delta 20.2$ (Me), $96.0(\mathrm{C}-7), 103.2(\mathrm{C}-3), 108.3$ (C-3a), $125.3,128.1,129.7(\mathrm{ArCH}), 131.9$ (ArC), 135.98 (C-2), 141.4 (C-7a), 156.5 (C-6), 160.4 (C=O); MS m/z 225 ( $\mathrm{M}^{+}, 100 \%$ ), 197 (35) (Calc. for $\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{NO}_{2}$ : C, 74.65; H, 4.92; N, 6.22. Found: C, $74.39 ; \mathrm{H}, 5.01 ; \mathrm{N}, 6.12 \%)$.
6-Methyl-4-( 1-morpholino-2-phenylethylideneamino)-2 H -pyran-2-one 5a. White plates from $\operatorname{Pr}_{2}{ }_{2} \mathrm{O} ; \mathrm{mp} 135^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 1705(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 2.15(3 \mathrm{H}$, s , Me), 3.45-3.47 ( $8 \mathrm{H}, 2 \mathrm{~m}$, morpholino), $3.80\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}\right), 5.29$ $(1 \mathrm{H}, \mathrm{d}, J 1.7,3-\mathrm{H}), 5.66(1 \mathrm{H}, \mathrm{d}, J 1.7,5-\mathrm{H}), 7.10-7.38(5 \mathrm{H}, \mathrm{m}$, ArH) ; ${ }^{13} \mathrm{C}$ NMR ( $\left.75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 20.3(\mathrm{Me}), 34.7\left(\mathrm{CH}_{2}\right)$, $45.8\left(\mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 66.8\left(\mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 96.4(\mathrm{C}-3), 105.0(\mathrm{C}-5)$, 127.6, 128.0, $129.6(\mathrm{ArCH}), 135.3(\mathrm{ArC}), 156.8(\mathrm{~N}-\mathrm{C}=\mathrm{N}), 161.8$ (C-6), $165.1(\mathrm{C}-4), 165.4(\mathrm{C}=\mathrm{O})\left(\right.$ Calc. for $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{3}: \mathrm{C}, 69.21$; H, 6.45; N, 8.97. Found: C, 69.33; H, 6.53; N, 8.88\%).

6-Methyl-2-p-tolylpyrano[4,3-b]pyrrol-4(1H)-one 4b. White plates from $\operatorname{Pr}^{\mathrm{i}}{ }_{2} \mathrm{O} ; \mathrm{mp} 268^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 3150(\mathrm{NH}), 1680$ $(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $\left.200 \mathrm{MHz} ; \mathrm{DMSO}-d_{6}\right) \delta 2.27(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Me}), 2.32(3 \mathrm{H}, \mathrm{s}, p$-tolyl), $6.48(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 6.93(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H})$, 7.24 and $7.66(4 \mathrm{H}, \mathrm{dd}, \mathrm{AB}$ system, $J 8.4, \mathrm{ArH}), 12.10(1 \mathrm{H}$, br s, NH exchangeable); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz} ;$ DMSO- $d_{6}$ ) $\delta 19.6$ (Me), 20.9 (Me), 95.4 (C-7), 101.8 (C-3), 107.7 (C-3a), 124.6, $129.7(\mathrm{ArCH}), 128.6(\mathrm{ArC}), 135.5(\mathrm{C}-2), 136.9(\mathrm{ArC}), 140.6$ (C-7a), $155.6(\mathrm{C}-6), 159.8(\mathrm{C}=\mathrm{O})$ (Calc. for $\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{NO}_{2}: \mathrm{C}$, 75.30 ; H, 5.48; N, 5.85. Found: C, 75.06; H, 5.57; N, 5.63\%).

6-Methyl-4-( 1 -morpholino-2-p-tolylethylideneamino)-2 H -pyran-2-one 5 b. White plates from $\mathrm{Pr}_{2}^{\mathrm{i}} \mathrm{O}$; mp $191{ }^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 2.15(3 \mathrm{H}$, s , Me), $2.32(3 \mathrm{H}, \mathrm{s}, p$-tolyl), $3.44-3.57(8 \mathrm{H}, \mathrm{m}$, morpholine), $3.74\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}\right), 5.28(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 5.65(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}), 6.99$ and $7.12(4 \mathrm{H}, \mathrm{dd}, \mathrm{AB}$ system, $J 8.4, \mathrm{ArH}) ;{ }^{13} \mathrm{C}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $\delta 19.9(\mathrm{Me}), 21.0(\mathrm{Me}), 33.9\left(\mathrm{CH}_{2}\right), 45.4\left(\mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 66.5$ $\left(\mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 96.2(\mathrm{C}-3), 104.7(\mathrm{C}-5), 127.5,129.9(\mathrm{ArCH})$, 131.8 (ArC), 136.8 ( ArC ), $156.6(\mathrm{~N}-\mathrm{C}=\mathrm{N}), 161.4$ (C-6), 164.8 (C-4), $165.0(\mathrm{C}=\mathrm{O})\left(\right.$ Calc. for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{3}: \mathrm{C}, 69.92 ; \mathrm{H}, 6.79$; N, 8.58. Found: C, 70.20; H, 6.72; N, 8.45\%).

2-(4-Methoxyphenyl)-6-methylpyrano[4,3-b]pyrrol-4(1H)one $4 c$. Cream plates from $\mathrm{CH}_{2} \mathrm{Cl}_{2} ; \operatorname{mp~} 242{ }^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }}$ $3230(\mathrm{NH}), 1670(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz} ; \mathrm{DMSO}-d_{6}$ )
$\delta 2.28(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.80(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 6.48(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 6.87$ $(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 7.02$ and $7.71(4 \mathrm{H}, \mathrm{dd}, \mathrm{AB}$ system, $J 8.7$, ArH), 12.05 ( $1 \mathrm{H}, \mathrm{br}$ s, NH exchangeable); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz ; DMSO-d $d_{6}$ ) $\delta 19.6$ (Me), 55.4 (OMe), 95.4 (C-7), 101.1 (C-3), 107.7 (C-3a), 114.6, 126.1 (ArCH), 124.2 (ArC), 135.5 (C-2), 140.49 (C-7a), 155.5 (C-6), 158.9 (OMe), 159.9 (C=O) (Calc. for $\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{NO}_{3}$ : C, 70.58; H, 5.13; N, 5.49. Found: C, 70.37; H, 4.96; N, 5.35\%).

4-[2-(4-Methoxyphenyl)-1-morpholinoethylideneamino]-6-methyl-2H-pyran-2-one 5 c. Orange plates from $\operatorname{Pr}_{2}{ }_{2} \mathrm{O} ; \mathrm{mp}$ $106{ }^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 200 MHz ; $\left.\mathrm{CDCl}_{3}\right) \delta 2.16(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.40-3.50\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 3.50-$ $3.60\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 3.72\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}\right), 3.80(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, $5.28(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 5.65(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}), 6.85$ and $7.03(4 \mathrm{H}, \mathrm{dd}, \mathrm{AB}$ system, $J$ 8.7, ArH) (Calc. for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{4}$ : C, $66.65 ; \mathrm{H}, 6.48$; N, 8.18. Found: C, 66.88; H, 6.71; N, 6.22\%).

2-(4-Chlorophenyl)-6-methylpyrano[4,3-b]pyrrol-4(1 H)-one $4 d$. White plates from. $\mathrm{CH}_{2} \mathrm{Cl}_{2} ; \mathrm{mp} 291^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }}$ $3230(\mathrm{NH}), 1670(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz} ;$ DMSO- $d_{6}$ ) $\delta 2.28(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 6.50(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 7.07(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 7.50$ and $7.80(4 \mathrm{H}$, dd, AB system, $J 8.5, \mathrm{ArH}), 12.22(1 \mathrm{H}, \mathrm{br}$ s, NH exchangeable); ${ }^{13} \mathrm{C}$ NMR ( 50 MHz DMSO- $d_{6}$ ) $\delta 19.6$ (Me), 95.3 (C-7), 103.3 (C-3), 107.8 (C-3a), 126.3, 129.1 (ArCH), 130.3 and $131.9(2 \times \mathrm{ArC}), 134.2(\mathrm{C}-2), 141.0(\mathrm{C}-7 \mathrm{a}), 156.1$ (C-6), $159.7(\mathrm{C}=\mathrm{O})\left(\right.$ Calc. for $\mathrm{C}_{14} \mathrm{H}_{10} \mathrm{ClNO}_{2}$ : $\mathrm{C}, 64.75 ; \mathrm{H}, 3.88$; N, 5.39. Found: C, $64.54 ; \mathrm{H}, 3.83$; N, $5.31 \%$ ).

4-[2-(4-Chlorophenyl)-1-morpholinoethylideneamino]-6-
methyl-2H-pyran-2-one 5d. Pale yellow plates from MeOH ; mp $158{ }^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 1690(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz ; $\left.\mathrm{CDCl}_{3}\right) \delta 2.16(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.38-3.50\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 3.50-$ $3.63\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 3.76\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}\right), 5.24(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H})$, $5.64(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}), 7.06$ and $7.31(4 \mathrm{H}, \mathrm{dd}, \mathrm{AB}$ system, $J 8.4, \mathrm{ArH})$ (Calc. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{ClN}_{2} \mathrm{O}_{3}$ : C, 62.34; H, 5.52; N, 8.08. Found: C, $62.08 ; \mathrm{H}, 5.43$; N, $7.95 \%$ ).
6-Methyl-4-( 1-morpholinopropylideneamino)-2H-pyran-2one 5e. White plates from $\operatorname{Pr}_{2}^{\mathrm{i}} \mathrm{O}$; mp $85-86^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $\delta 1.11(3 \mathrm{H}$, $\mathrm{t}, J 7.6, \mathrm{Me}), 2.19(3 \mathrm{H}, \mathrm{s}$, pyranone-Me), 2.36 ( $2 \mathrm{H}, \mathrm{dd}, J 7.6$, $\left.\mathrm{CH}_{2}\right), 3.45-3.51$ and $3.70-3.76(8 \mathrm{H}, 2 \mathrm{~m}$, morpholine), $5.24(1 \mathrm{H}$, d, $J 1.7,3-\mathrm{H}$ ), $5.62(1 \mathrm{H}, \mathrm{d}, J 1.7,5-\mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz ; DMSO- $d_{6}$ ) $\delta 12.4\left(\right.$ Me linked to $\left.\mathrm{CH}_{2}\right), 19.6(6-\mathrm{Me}), 21.7\left(\mathrm{CH}_{2}\right)$, $45.2\left(\mathrm{CH}_{2} \mathrm{NCH}_{2}\right), 66.2\left(\mathrm{CH}_{2} \mathrm{OCH}_{2}\right), 95.0(\mathrm{C}-3), 104.5(\mathrm{C}-5)$, $160.6(\mathrm{~N}-\mathrm{C}=\mathrm{N})$, 161.5 (C-6), 163.7 (C-4), 165.39 (C=O) (Calc. for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{3}$ : C, 62.38; H, 7.25; N, 11.19. Found: C, 62.13; H, 7.46; N, 11.03\%).

2-(4-Bromophenyl)-6-methylpyrano[4,3-b]pyrrol-4(1H)-one 4f. White plates from $\mathrm{CH}_{2} \mathrm{Cl}_{2} ; \mathrm{mp}>300^{\circ} \mathrm{C}$ (decomp.); IR (Nujol) $v_{\text {max }} 3200(\mathrm{NH}), 1660(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 200 MHz ; DMSO-d $d_{6}$ ) $2.27(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 6.5(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 7.07(1 \mathrm{H}, \mathrm{s}$, $3-\mathrm{H}), 7.60-7.80(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 12.25(1 \mathrm{H}, \mathrm{br} \mathrm{s}$, NH exchangeable); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz} ;$ DMSO- $d_{6}$ ) $\delta 19.32$ (Me), 95.04 (C-7), 103.00 (C-3), 107.44 (C-3a), 126.23, 131.67 (ArCH), 120.02 and $130.26(2 \times \mathrm{ArC}), 133.84$ (C-2), 140.72 (C-7a), 155.80 (C-6), $159.38(\mathrm{C}=\mathrm{O})\left(\right.$ Calc. for $\mathrm{C}_{14} \mathrm{H}_{10} \mathrm{BrNO}_{2}: \mathrm{C}, 55.29 ; \mathrm{H}, 3.31$; N, 4.61. Found: C, 55.03 ; H, 3.34; N, $4.55 \%$ ).

4-[2-(4-Bromophenyl)-1-morpholinoethylideneamino]-6-methyl-2H-pyran-2-one 5 f. Chestnut plates from $\mathrm{EtOH} ; \mathrm{mp}$ $158{ }^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 1695(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( 200 MHz ; $\left.\mathrm{CDCl}_{3}\right) \delta 2.08(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 3.35-3.55(8 \mathrm{H}, \mathrm{m}$, morpholine), 3.69 $\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}\right), 5.15(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 5.58(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}), 6.94$ and 7.37 $\left(4 \mathrm{H}\right.$, dd, AB system, $J 8.4, \mathrm{ArH}$ ) (Calc. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{BrN}_{2} \mathrm{O}_{3}$ : C, 55.38 ; H, 4.91; N, 7.18. Found: C, 55.17; H, 4.98; N, 6.93\%).
2-(4-Fluorophenyl)-6-methylpyrano[4,3-b]pyrrol-4(1H)-one $4 g$. White plates from $\mathrm{CH}_{2} \mathrm{Cl}_{2} ; \mathrm{mp} 248^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 3100$ (NH), $1680(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz} ;$ DMSO- $d_{6}$ ) $\delta 2.27$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 6.49(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 7.00(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 7.20-7.40$ and 7.85-7.95 ( $4 \mathrm{H}, 2 \mathrm{~m}, \mathrm{ArH}$ ), 12.18 ( 1 H, br s, NH exchangeable); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz} ;$ DMSO- $d_{6}$ ) $\delta 19.31$ (Me), 95.04 (C-7), 102.25 (C-3), 107.35 (C-3a), 115.54 and 115.97 (ArCH ortho to F, J 21.6), 126.33 and 126.48 ( ArCH meta to F, $J 7.6$ ), 127.70
$(\operatorname{ArC}), 134.11(\mathrm{C}-2), 140.46(\mathrm{C}-7 \mathrm{a}), 155.62(\mathrm{C}-6), 158.9$ and 159.46 (CF, $J$ 245.0), $159.46(\mathrm{C}=\mathrm{O})$ (Calc. for $\mathrm{C}_{14} \mathrm{H}_{10} \mathrm{FNO}_{2}$ : C, 69.12; H, 4.15; N, 5.76. Found: C, 68.97; H, 4.05; N, 5.64\%). 4-[2-(4-Fluorophenyl)-1-morpholinoethylideneamino]-6-methyl-2H-pyran-2-one 5g. White needles from $\mathrm{Pr}^{\mathrm{i}} \mathrm{OH} ; \mathrm{mp}$ $126^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 1690(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 200 MHz ; $\left.\mathrm{CDCl}_{3}\right) \delta 2.13\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 3.40-3.58(8 \mathrm{H}, \mathrm{m}$, morpholine), $3.75\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}\right), 5.22(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}), 5.63(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}), 6.95-7.15$ ( $4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ) (Calc. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{FN}_{2} \mathrm{O}_{3}: \mathrm{C}, 65.43 ; \mathrm{H}, 5.80$; N, 8.48. Found: C, 65.29; H, 5.87; N, 8.33\%).

## Method B

A solution of a dihydrotriazole $3 \mathbf{a - g}(10 \mathrm{mmol})$ in propan-1-ol $(60 \mathrm{ml})$ was boiled under reflux, progress of the reaction being followed by TLC. After disappearance of the starting material the solvent was removed in vacuo, a small amount of residue was dissolved in DMSO- $d_{6}$ and the ratio of $\mathbf{4}$ and 5 was determined by ${ }^{1} \mathrm{H}$ NMR analysis (see Table 1). Compounds $\mathbf{4}$ and $\mathbf{5}$ were isolated after column chromatography (cyclohexane-ethyl acetate $3: 7$ ). The dihydrotriazole 3 e yielded only the amidine $\mathbf{5 e}$, which was crystallized. Analytical data were in agreement with those previously reported. The reaction times and isolated yields of compounds are collected in Table 2.

## Method C. Effect of $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathbf{O}$ with respect to pyrolysis of dihydrotriazoles 3a-e

$\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}(1.9 \mathrm{ml}, 15 \mathrm{mmol})$ in dry toluene ( 30 ml ) was added to a stirred solution of a dihydrotriazole 3a-e $(4.2 \mathrm{~g}, 15 \mathrm{mmol})$ in dry toluene $(25 \mathrm{ml})$ at rt . The resulting mixture was refluxed for the same time used in the toluene thermal decomposition, after which the reaction was quenched with 10 ml of saturated aq. $\mathrm{NaHCO}_{3}(10 \mathrm{ml})$. The organic phase was separated, and the aqueous phase was extracted with ethyl acetate $(4 \times 50 \mathrm{ml})$. The combined organic phases were washed with brine and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered, and evaporated to dryness. Ratio of $\mathbf{4}$ and 5 determined by ${ }^{1} \mathrm{H}$ NMR analysis (DMSO- $d_{6}$ ) (see Table 1) of the residue. The crude mixtures containing $\mathbf{4 a - g}$ and $\mathbf{5 a - g}$ were chromatographed (cyclohexane-ethyl acetate $2: 8$ ) to afford first the pyrano[4,3-b]pyrrol-4-ones 4 and then the amidines 5. For analytical data of the products previously obtained see above. The reaction times and isolated yields of compounds are collected in Table 2.

2,6-Dimethylpyrano[4,3-b]pyrrol-4( $\mathbf{1 H}$ )-one 4e. White needles from $\operatorname{Pr}^{\mathrm{i}} \mathrm{O} ; \mathrm{mp} 157^{\circ} \mathrm{C}$; IR (Nujol) $v_{\text {max }} 3090(\mathrm{NH}), 1670(\mathrm{C}=\mathrm{O})$ $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $\delta 2.26(3 \mathrm{H}, \mathrm{s}, 6-\mathrm{Me}), 2.34$ $(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{Me}), 6.27(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 6.33(1 \mathrm{H}, \mathrm{s}, 3 \mathrm{H}), 9.20(1 \mathrm{H}, \mathrm{br} \mathrm{s}$, NH exchangeable); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $\delta 13.5$ (2-Me), 20.2 (6-Me), 96.3 (C-7), 103.4 (C-3), 107.8 (C-3a), 133.1 (C-2), $140.64(\mathrm{C}-7 \mathrm{a}), 155.2(\mathrm{C}-6), 162.4(\mathrm{C}=\mathrm{O})$ (Calc. for $\mathrm{C}_{9} \mathrm{H}_{9} \mathrm{NO}_{2}$ : C, $66.25 ; \mathrm{H}, 5.56 ; \mathrm{N} ; 8.58$. Found: C, 66.06 ; H, 5.74; N; 8.73\%).

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